

A Process Development FOR Extraction of Marker Compound from *EMBLICA OFFICINALIS* FRUITSaroj Yadav^{*1}, Anoop Kumar², Om Prakash³^{*1}Institute of Pharmaceutical Sciences and Research Center, Bhagwant University, Ajmer Rajasthan, India²Department of Pharmaceutical Technology, MEIT Meerut, India³R&D Center, AVA CHCPL India

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Abstract

The present study is done to examine and optimize the process of enzyme-assisted extraction of Gallic acid from the dried fruit of *Embllica officinalis*. The aqueous extraction yielded 12.09% of Gallic acid. A selective extraction process after the treatment with enzymes is proposed by using 30% (v/v) methanol which releases up to 15.68% of the Gallic acid, present in the dried fruit. The optimal conditions were as follows: pH value was 4.5, concentration of cellulase solution was 2.5 mg/mL, incubation time was 8 h, incubation temperature was 50 °C and solid:solvent ratio was 1:8. Enzyme-assisted extraction was proven to be highly efficient in the process development for extraction of marker compounds from herbal drugs and could be used in making enriched herbal extracts of product on a large scale.

Keywords: *Embllica officinalis*, Total Gallic acid, enzymatic extraction**Introduction**

Embllica officinalis is effective in the treatment of amlapitta viz., peptic ulcer¹⁻². The fruits exhibit hypolipidaemic and antiatherosclerotic effects in the rabbits and rats³⁻⁴. The extract of amla also has antimicrobial properties⁵⁻⁶. Amla is an antioxidant with free radical scavenging properties⁷. Hepatoprotective⁸, adaptogenic⁹, antimutagenic¹⁰, cytoprotective and antitumor¹¹ antifungal¹², were also exhibited by Amla. The Gallic acid is the basis for the quality control of *Amla* and other plant-derived drugs from the herb.

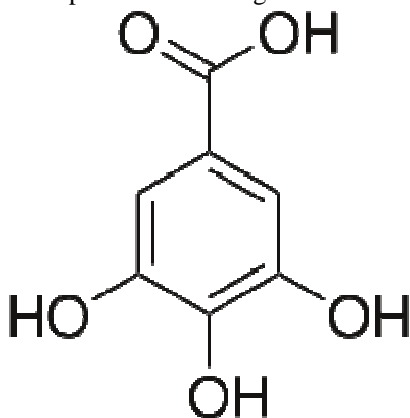


Fig. 1. Chemical structure of Gallic acid.

Although many reports about extracting Gallic acid by using different extraction methods from different plants have been published [13], some disadvantages still exist such as large organic solvent consumption, and low efficiency. Therefore, high efficient extraction method of Gallic acid represents a hot spot in *Embllica officinalis* research. Enzyme-assisted extraction is a method applied to the study of secondary metabolites releasing from biogenic materials. It possess the advantages of environmental- friendship, high efficiency and easy operation process. It has been represented as an alternative way for extracting marker compound from the herbal drug. Hydrolytic enzymes including cellulase, beta-glucosidase and pectinase, which are commonly used in extraction [14-16], can interact on cell wall, break down its structural integrity so as to increase the releasing of Gallic acid notably.

The main aim of the present study is to examine and optimize the process of enzyme-assisted extraction of Gallic acid from the Fruits of *Embllica officinalis*. For this purpose, the selection of enzyme type, pH and the concentration of enzyme solution, incubation time and temperature were studied, in order to obtain high yields of above natural products economically and environmental friendly.

Material and Methods

Plant material

The dry Amla was collected and ground into fine powder using a high-speed blender. The dry, ground Amla was packed in a plastic bag, sealed and kept in the refrigerator (5°C) until used.

Chemicals and reagents

Gallic acid, and Cellulase were provided by Radiant Research Pvt. Ltd as gift sample. Methanol of analytical grade were purchased from Rankem Ltd. and double-distilled water was used in all experiments

Enzyme-assisted extraction and pretreatment

Cellulase was quantified accurately and dispersed in deionized water to obtain enzyme solutions of certain concentrations (0.25-4 mg/mL). 100 g dry powder was added to the enzymatic solution and adjusted to certain pH (3.5-7.0) with 0.1 M HCl solution and shaken on a flat-bed orbital shaker for a period of time (1-10 hr) at certain temperature (30-55°C). After the treatment fulfilled, the extract was filtered through Whatmann filter paper no 1. Filtrate collected was concentrated in vacuo (55°C) in a rotary evaporator and analyzed by spectrophotometer. All the experiments were performed in triplicate.

Quantification of Gallic acid

Quantitative determination of total Gallic acid content in each sample of *Amla* was performed by the described method [17]. Gallic acid content was calculated using a area under standard curve. Analysis of each sample was done in triplicate. For preparation of standard solution, standard Gallic acid (2.00 mg) was accurately weighed and transferred to a 5-ml volumetric flask. Methanol was added and adjusted to a final concentration of 400 µg/ml. From this solution, concentrations of 0.8, 1.6, 2.0, 2.4 and 3.2 µg/ml were prepared and used for preparation of the calibration curve. For preparation of sample solution from *Amla*, the extract (300.00 mg) of each sample was separately transferred to a 10-ml volumetric flask. One milliliter of the clear supernatant liquid was transferred and diluted with methanol to 25 ml volume. This solution (1 ml) was then transferred to a 50-ml volumetric flask, and diluted to volume with methanol and % Gallic acid was calculated.

Results and Discussion

Cellulase catalyzes the breakdown of cellulose into glucose, cellobiose and higher glucose polymers; therefore, it is used to extract Gallic acid from *Amla* Fruit.

Effect of pH value of enzyme solution

It has been reported that the activity of cellulase can be influenced by pH very much, and it is believed that it works better with pH<7 [18]. The effect of pH was studied in this experiment in order to pick out the proper pH value which would make the cellulase work best. Fig.2 shows the effect of pH on the extraction yields of the Gallic acid. It can be observed that the yields of Gallic acid varied unregularly with different pH value. The yields of Gallic acid achieved the maximum at pH 4.5.

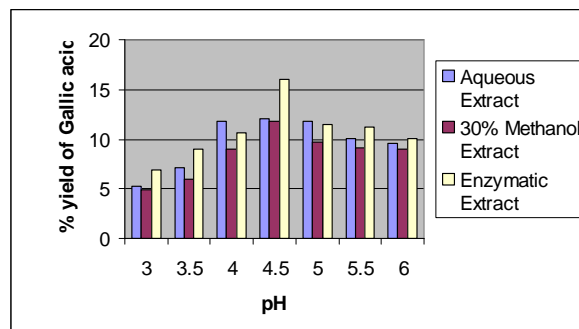


Fig. 2. Effect of pH on the yield of Gallic acid

Effect of enzyme concentration

The effect of concentration of cellulase on the extraction yields of Gallic acid was studied and the results are shown in Fig. 3. According to the results, it is obvious that with the increasing of cellulase concentration, the yields of these Gallic acid increased gradually until 2.5 mg/mL. Comparing with the yields of Gallic acid at the concentration of 2.5 mg/mL, 4.0 mg/mL did not show distinct advantage. Considering the economic influence, 2.5 mg/mL was selected for the pretreatment of the extraction process.

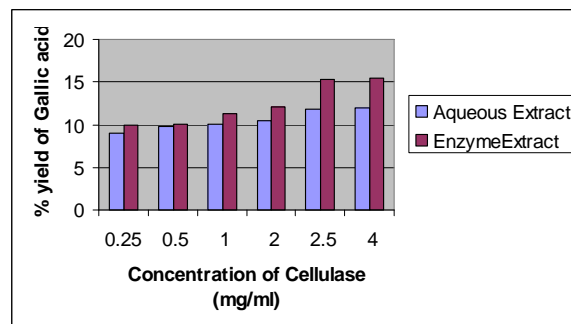


Fig. 3. Effect of conc. of Cellulase on yield of Gallic acid

Effect of incubation time

Fig. 4 showed the results of the effect of cellulase incubation time on the extraction yields of Gallic acid. The yields of these Gallic acid increased

notably along with the extending of incubation time. The yields of Gallic acid reached the peak (15.68 mg/g) at 8 h. And the yields began to decrease in additional time. Thus, 8 h was considered to be enough for cellulase to catalyze the hydrolysis of cell wall in Fruit of *Emblica officinalis*.

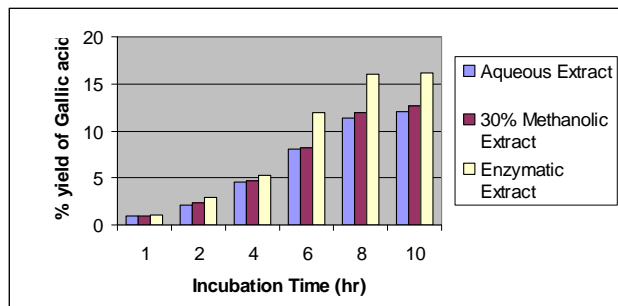


Fig. 4. Effect of Incubation time on the yield of Gallic acid

Solid Solvent Ratio

Different solid: solvent ratios ranging from 1:2 to 1: 12 were studied and the optimum ratio for the extraction of Gallic acid was found to be 1:8 g/ml (Fig 5).

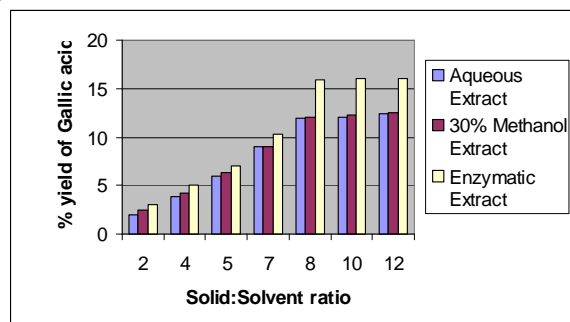


Fig. 5. Effect of Solid:Solvent ratio on the yield of Gallic acid

Effect of temperature on enzyme activity

The study of the thermal effect on the extraction yields was also carried out in this work. The results are presented in Fig.6. The yields of Gallic acid varied with the change in temperature. With the increase in temperature, the yields of Gallic acid increased gradually until 50 °C. The yields of Gallic acid increased upto 15.68 mg/ml. Therefore, 50 °C was chosen for cellulase incubation temperature in this assay.

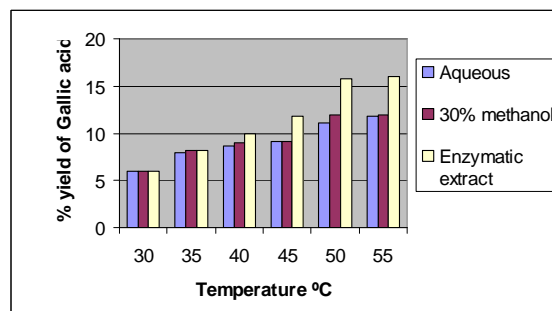


Fig. 6. Effect of Temperature on the yield of Gallic acid

Statistical analysis

All results were subjected to statistical analyses. Mean values of all data were obtained from triplicate experiment.

Conclusion

Enzyme-assisted extraction Gallic acid, from Fruits of *Emblica officinalis* was carried out in present study. The effect of hydrolytic enzyme was studied and it was proved that cellulase at a concentration of 2.5mg/ml to be most effective for extracting Gallic acid from Fruit of *Emblica officinalis*. As per the economic effect, cellulase was chosen for the treatment of the Fruit. The extraction conditions including pH and the concentration of cellulase solution, solid:solvent ratio, incubation time and incubation temperature were optimized. Results showed that all these factors were important for the extraction of Gallic acid. The optimal conditions were as following: pH value was 4.5, concentration of cellulase solution was 2.5 mg/mL, incubation time was 8 h, incubation temperature was 50 °C and solid:solvent ratio was 1:8 . Pass through the treatment by cellulase, the contents of Gallic acid were 1.23-fold, of those in the control which showed that cellulase destroy the structures of plant cells and results in higher extraction yields of Gallic acid. Enzyme-assisted extraction may provide a feasible way for the extraction of Gallic acid from *Emblica officinalis* and other species of Amla, it has the advantages of environment friendship, lower cost, easy operation and higher efficiency, and it is promising for industry application broadly.

References

- [1] Al-Rehailya, A.J., Al-Howirinya, T.S., Al-Sohaibanib, M.O., and Rafatullaha, S. (2002). Gastroprotective effects of 'Amla' *Emblica officinalis* on in vivo test models in rats. *Phytomedicine* 9(6), 515-522.
- [2] Sairam, K., Rao, C.V., Babu, M.D., Kumar, K.V., Agrawal, V.K., and Goel, R.K. (2002). Antiulcerogenic effect of methanolic extract of *Emblica officinalis*: an experimental study. *J Ethnopharmacol* 82(1), 1-9.
- [3] Anil, L., Vijayalakshmi N.R. (2003). Antioxidant action of flavonoids from *Mangifera indica* and *Emblica officinalis* in hypercholesterolemic rats. *Food Chemistry* 83 569-574.
- [4] Yokozawa T., Kim H.Y., Kim H.J., Okubo T., Chu D.C., Juneja L.R. (2007). Amla (*Emblica officinalis* Gaertn.) prevents dyslipidaemia and oxidative stress in the ageing process. *Br J Nutr.* 97(6):1187-95.
- [5] Ahmed I., Mehmood Z. and Mohammad F. (1998). Screening of some Indian medicinal plants for their antimicrobial properties. *J. Ethnopharmacol* 62, 183-93.
- [6] Saeed S., Tariq P. (2007). Antibacterial activities of *Emblica officinalis* and *Coriandrum sativum* against Gram negative urinary pathogens. *Pak J Pharm Sci.* 20(1), 32-35.
- [7] Bhattacharya, A., Chatterjee, A., Ghosal, S., Bhattacharya, S. K. (1999). Antioxidant activity of active tannoid principles of *Emblica officinalis* (amla). *Indian J. Exp. Biol.*, 37: 676-680.
- [8] Mir AI, Kumar B, Tasduq SA, Gupta DK, Bhardwaj S, Johri RK. (2007). Reversal of hepatotoxin-induced pre-fibrogenic events by *Emblica officinalis*-a histological study. *Indian J Exp Biol.* 45(7), 626-629.
- [9] Rege, N. N., Thatte, U. M., Dahanukar, S. A. (1999) Adaptogenic properties of six rasayana herbs used in Ayurvedic medicine. *Phytother. Res.*, 13: 275-91.
- [10] Kaur S., Arora S., Kaur K. (2002). The in vitro antimutagenic activity of Triphala- an Indina herbal drug. *Food Chem Toxicol* 40, 527-534.
- [11] Jose J.K., Kuttan G., Kuttan R. (2001). Antitumour activity of *Emblica officinalis* J. *Ethnopharmacol* 75, 65-69.
- [12] Dutta B. K. Rahman I., Das T.K. (1998). Antifungal activity of Indian Plant extracts. *Mycoses* 41, 535-536.
- [13] Magdalena Karamae, Agnieszka Kosińska, Ronald B. Pegg. Content of gallic acid in selected plant extracts. *Pol. J. Food Nutr. Sci.* 2006, Vol. 15/56, No 1, pp. 55-58
- [14] Inci Çinar., *Process Biochem* 2005; 40: 945-949.
- [15] Y.J. Kim, D.K. Kim, O.K. Chun, *J. Agric. Food Chem* 2005; 53: 9560-9565.
- [16] R.M. Wilkins, W.W. Widmer, K. Grohmann, *Bioresource Technol* 2007; 98: 1596-1601. Om Prakash, Singh R.M., Mathur S.C., Singh
- [17] G.N. Quantification of Gallic acid by HPLC and Antioxidant activity of Amla fruits. *J. Pharmaceutical Research*, 2007, 6 (3), 161-162.